

Stereoselective total synthesis of furofuran lignans through dianion aldol condensation

Jae-Chul Jung,^a Ju-Cheun Kim,^a Hyung-In Moon^b and Oee-Sook Park^{c,*}

^aDepartment of Medicinal Chemistry, School of Pharmacy, University of Mississippi, PO Box 1848, University, MS 38677-1848, USA

^bNational Center of Natural Product Research, School of Pharmacy, University of Mississippi, PO Box 1848, University, MS 38677-1848, USA

^cDepartment of Chemistry, Institute for Basic Sciences, College of Natural Sciences, Chungbuk National University, Cheongju 361-763, Chungbuk, South Korea

Received 12 May 2006; revised 22 June 2006; accepted 23 June 2006

Available online 24 July 2006

Abstract—Highly stereoselective total synthesis of (+)-eudesmin, (+)-yangambin, (–)-eudesmin, and (–)-yangambin is described. This method is useful to generate the core skeleton of furofuran rings utilizing modification of Evans asymmetric aldol condensation. © 2006 Elsevier Ltd. All rights reserved.

1. Introduction

Furofuran lignans have stimulated significant interest due to their wide range of intriguing biological activities¹ such as antitumor,² antimitotic,³ antiviral,⁴ antioxidant,⁵ antihypertensive,⁶ inhibition of platelet-activating factor (PAF),⁷ and Ca²⁺ channels,⁸ cAMP phosphodiesterase inhibitory,⁹ sodium selective diuretic properties,¹⁰ and microsomal monooxygenases inhibitory effects for insects.¹¹ Several elegant synthetic approaches have been reported¹² in the literature, most of these methods are substrate specific providing an entry to one particular stereochemical series and only a few approaches are applicable to the stereocontrolled synthesis of exo–exo furofuran (±)-sesamin, (±)-eudesmin, (±)-yangambin, and (±)-epiasarinin (Fig. 1).^{12f,g}

We recently reported¹³ an efficient asymmetric synthesis of (+)-sesamin and (–)-sesamin. Key steps include highly stereoselective aldol condensation of piperonal with the dianion of chiral auxiliary, followed by intramolecular ring cyclization of aldol product in high yield. In connection with our synthetic approaches, we have

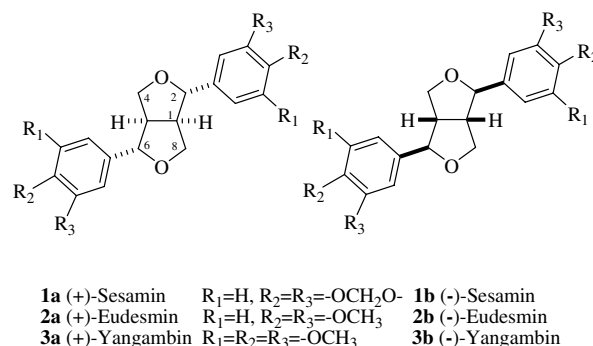


Figure 1. Structures of furofuran lignans.

been interested in the asymmetric total synthesis of lignans of the furofuran series.

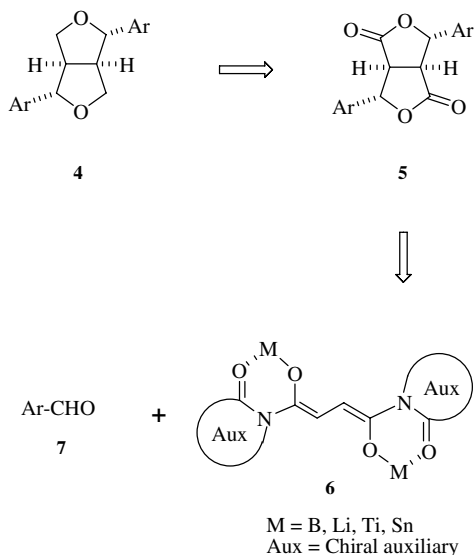
Herein, we described the efficient stereoselective total synthesis of (+)-eudesmin, (+)-yangambin, (–)-eudesmin, and (–)-yangambin by modification of asymmetric aldol condensation and intramolecular ring cyclization.

2. Results and discussion

The general features of our initial approach to the generation of the 2,6-diaryl-3,7-dioxabicyclo[3.3.0]octane skeleton of furofuran lignans are summarized in

Keywords: Furofuran lignans; (+)-Eudesmin; (+)-Yangambin; (–)-Eudesmin; (–)-Yangambin; Stereoselective aldol condensation; Intramolecular ring cyclization.

* Corresponding author. Tel.: +82 43 261 2283; fax: +82 43 267 2279; e-mail: ospark@cbnu.ac.kr



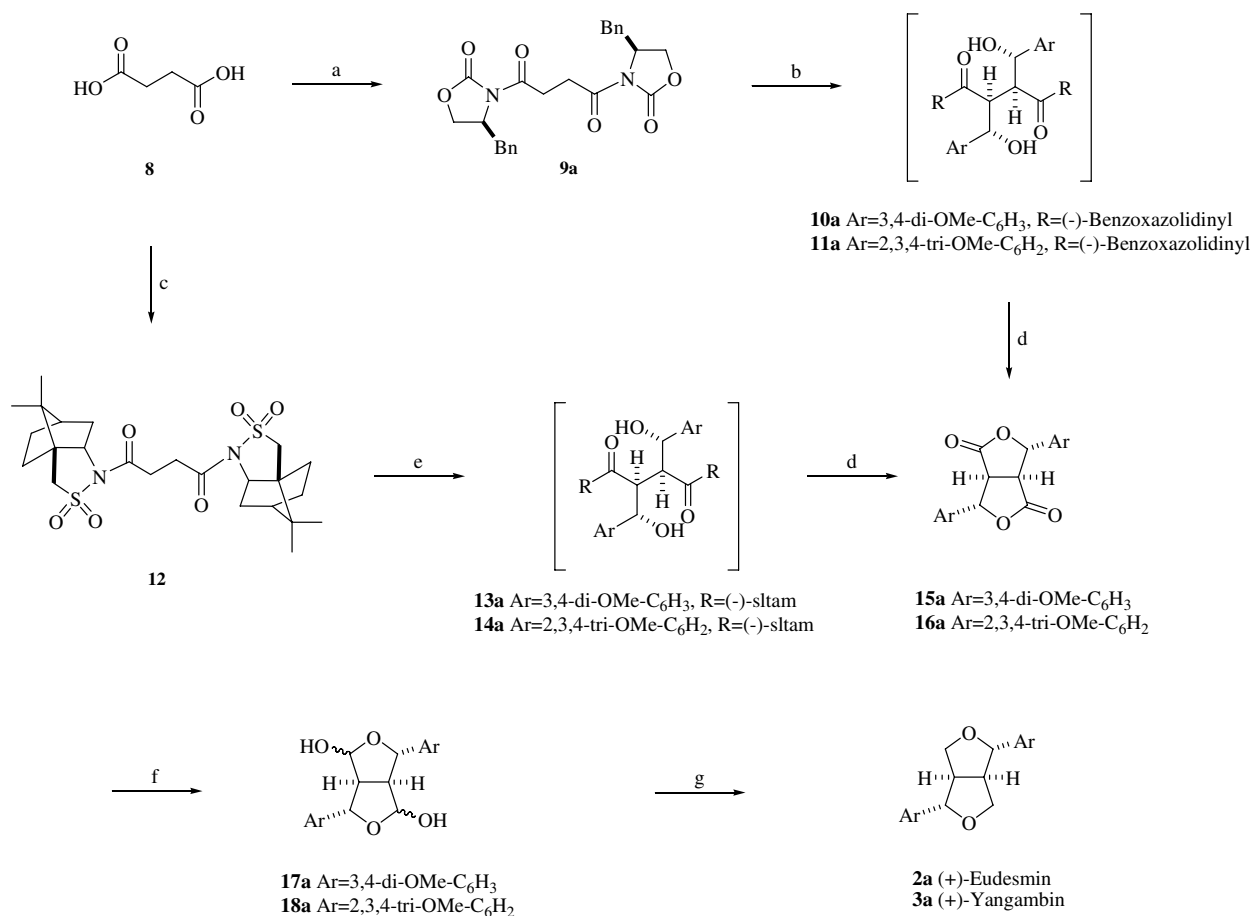
Scheme 1. Retrosynthetic analysis of eudesmin and yangambin.

Scheme 1. The chelated metal complex **6** (derived from **9a** or **12** with enolating agents) would be served as a key intermediate according to the retrosynthetic analysis

of the furofuran lignans shown in **Scheme 1**. A diastereoselective double aldol condensation of chelated metal complex **6** and aldehyde **7**, followed by intramolecular ring cyclization and double-reduction to complete the synthesis of furofuran lignans **4**.

The synthesis of (+)-eudesmin (**2a**) and (+)-yangambin (**3a**) was accomplished as shown in **Scheme 2**. Thus, succinic acid (**8**) was treated with thionyl chloride to give succinyl chloride, which was reacted with freshly prepared (*S*)-(-)-4-benzyl-2-oxazolidinone and (*S*)-(-)-2,10-camphorsultam as chiral auxiliaries [(*S*)-(-)-4-benzyl-2-oxazolidinone was prepared¹⁴ from (*S*)-phenylalaninol, diethyl carbonate, and sodium bicarbonate in 80% yield; (*S*)-(-)-2,10-camphorsultam was obtained from (+)-camphor-10-sulfonic acid with thionyl chloride, ammonium hydroxide, and sodium borohydride in a 65% three steps yield¹⁵] in the presence of *n*-BuLi (1.6 M soln, in hexane) at 0 °C to afford the required 1,4-bis[4-(*S*)-benzyl-2-oxazolidin-3-yl]butane-1,4-dione (**9a**) and (*S*)-(-)-sultam (**12**) in 85% and 88% yields, respectively.

The stereoselective aldol condensation of aldehydes (3,4-dimethoxybenzaldehyde, 2,3,4-trimethoxybenzaldehyde) and chiral auxiliaries **9a**, **12** into the dichloro-



Scheme 2. Reagents and conditions: (a) SOCl₂, CH₂Cl₂, reflux, 1 h; then (*S*)-(-)-4-benzyl-2-oxazolidinone, *n*-BuLi/THF, 0 °C, 2 h, 85%; (b) Bu₂BOTf, DIPEA/CH₂Cl₂, -78 °C, 3,4-dimethoxybenzaldehyde, 20 min; (c) SOCl₂, CH₂Cl₂, reflux, 1 h; then (*S*)-(-)-2,10-camphorsultam, *n*-BuLi/THF, 0 °C, 2 h, 88%; (d) KH₂PO₄ (aq), H₂O₂ (28%)/MeOH, 25 °C, 8 h, 84% for **15a**, 81% for **16a**; (e) Bu₂BOTf, DIPEA/CH₂Cl₂, -78 °C, 2,3,4-trimethoxybenzaldehyde, 20 min; (f) DIBAL-H/THF, -25 °C, 1 h, 42% for **17a**, 45% for **18a**; (g) Et₃SiH, BF₃·Et₂O/CH₂Cl₂, -78 °C, 1 h, 55% for **2a**, 58% for **3a**.

methane was performed with dibutylboron triflate at $-78\text{ }^{\circ}\text{C}$. The resultant dihydroxyl compounds **13a** and **14a** were unstable and we were not able to isolate them in sufficient purity or yield. Compounds **13a** and **14a** were subjected to intramolecular ring cyclization using KH_2PO_4 and H_2O_2 (28%) in MeOH to yield dilactone **15a** and **16a** in 84% and 81% two step yields, with 18:1, 20:1 diastereomeric selectivity, respectively. Conversion of **15a** and **16a** was achieved by reduction using DIBAL-H in THF, followed dehydroxylation of dilactols **17a** and **18a** to generate (+)-eudesmin (**2a**) and (+)-yangambin (**3a**) (Scheme 2).

The key step in our synthesis involved the stereoselective aldol coupling of enolizable substrates **9a** and **12** with aldehydes **7**. The primary goal in a reaction of this type is to obtain maximum diastereoselectivity. Therefore, we sought to determine the conditions necessary to achieve this goal by varying the enolizable substrate, and the reaction conditions for the aldol condensation. In our

original report,¹³ oxazolidinone **9a** was used as an enolizable substrate resulting in a 19:1 diastereomeric ratio of dilactones as (+)-, (–)-sesamin precursors in good yield. Having pure enolizable substrate **9a** and **12** in hand, we tested several enolating agents¹⁶ such as $\text{Et}_2\text{BOTf}/\text{CH}_2\text{Cl}_2$, $\text{Bu}_2\text{BOTf}/\text{CH}_2\text{Cl}_2$, $n\text{-BuLi}$, ZnCl_2/THF , LDA/THF , $\text{TiCl}_4/\text{CH}_2\text{Cl}_2$, $\text{TiCl}(\text{O}i\text{Pr})_3/\text{CH}_2\text{Cl}_2$, and $\text{SnOTf}/\text{CH}_2\text{Cl}_2$ under standard conditions.¹⁷ The best results were obtained with $\text{Bu}_2\text{BOTf}/\text{CH}_2\text{Cl}_2$ at $-78\text{ }^{\circ}\text{C}$ to $0\text{ }^{\circ}\text{C}$ for 1 h, regarding diastereoselectivity and yield (Table 1). Although stereoselectivity was enhanced, the reaction failed to go to completion. Aldol reactions with lithium, titanium, and tin failed to enhance the stereoselectivity. Also, the enhancement in diastereoselectivity was not sufficient to compensate for the low yields in both aldol condensation and intramolecular ring cyclization. We have found that our original reaction conditions using the chelated boron complex of chiral auxiliaries **9a** and **12**, followed by intramolecular ring cyclization with KH_2PO_4 and

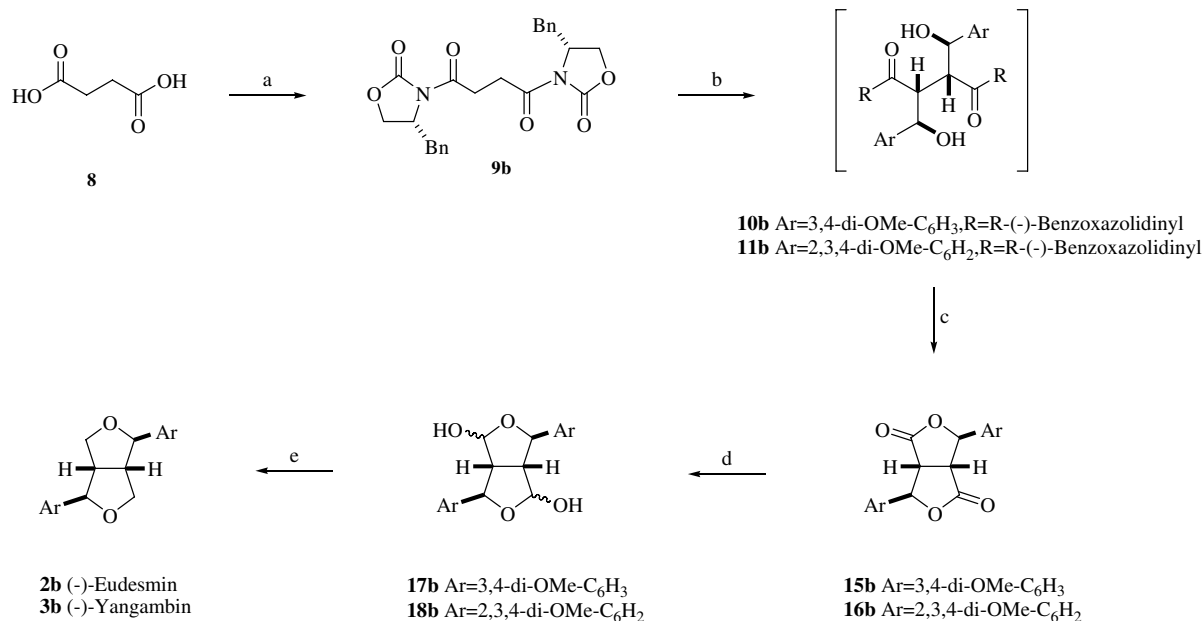
Table 1. Stereoselective aldol condensation and intramolecular ring cyclization of **10a** and **11a** to form dilactones **15a** and **16a**

Entry	Substrate	Enolization conditions	Temp ($^{\circ}\text{C}$)	Time (h)	Selectivity; yield (%)	
					15a	16a
1	9a or 12	Et_2BOTf , DIPEA/ CH_2Cl_2	-78 to 0	1	15:1 ^a (48) ^b	10:1 (33)
2		Bu_2BOTf , DIPEA/ CH_2Cl_2	-78 to 0	1	18:1 (61)	9:1 (28)
3		$n\text{-BuLi}$, ZnCl_2/THF	-40 to 0	1	20:1 (45)	25:1 (35)
4		LDA/THF	-78	0.5	5:1 (34)	10:1 (37)
5		$\text{TiCl}_4/\text{CH}_2\text{Cl}_2$	-78 to 0	2	10:1 (38)	— ^c
6		$\text{TiCl}(\text{O}i\text{Pr})_3/\text{CH}_2\text{Cl}_2$	-78 to 0	2	10:1 (30)	— ^c
7		$\text{Sn}(\text{OTf})_2/\text{CH}_2\text{Cl}_2$	-78 to 0	1	5:1 (22)	— ^c

^a Stereoselectivity was determined by HPLC analysis.

^b Yield was isolated yield.

^c No isolation.



Scheme 3. Reagents and conditions: (a) SOCl_2 , CH_2Cl_2 , reflux, 1 h; then (*S*)-(-)-4-benzyl-2-oxazolidinone, $n\text{-BuLi}/\text{THF}$, $0\text{ }^{\circ}\text{C}$, 2 h, 86%; (b) Bu_2BOTf , DIPEA/ CH_2Cl_2 , $-78\text{ }^{\circ}\text{C}$, 3,4-dimethoxybenzaldehyde or 2,3,4-trimethoxybenzaldehyde, 20 min; (c) KH_2PO_4 (aq), H_2O_2 (28%)/MeOH, $25\text{ }^{\circ}\text{C}$, 8 h, 80% for **15b**, 83% for **16b**; (d) DIBAL-H/THF, $-25\text{ }^{\circ}\text{C}$, 1 h, 42% for **17b**, 45% for **18b**; (e) Et_3SiH , $\text{BF}_3\cdot\text{Et}_2\text{O}/\text{CH}_2\text{Cl}_2$, $-78\text{ }^{\circ}\text{C}$, 1 h, 51% for **2b**, 50% for **3b**.

H₂O₂ (28%) provided acceptable selectivity and yield for preparation of the furofuran lignans **4**.

Likewise, treatment of oxazolidinone **9b** with aldehydes (3,4-dimethoxybenzaldehyde, 2,3,4-trimethoxybenzaldehyde) in the presence of Bu₂BOTf in dry dichloromethane gave diastereomeric boron complex mixtures, which were subjected to intramolecular ring cyclization with KH₂PO₄ and H₂O₂ (28%) in MeOH to afford dilactones **15b** and **16b** in 80% and 83% yields, with 19:1 diastereomeric selectivity, respectively. Reduction of dilactones **15b** and **16b** with DIBAL-H in THF gave dilactols **17b** and **18b** in good yields, which were treated with Et₃SiH and BF₃·Et₂O in dichloromethane to give (–)-eudesmin (**2b**) and (–)-yangambin (**3b**) (Scheme 3). The properties of **2b** and **3b** were identical to the reported spectral and physical data for these compounds.^{12f,1}

In conclusion, we have developed a method for the efficient synthesis of (+)-, (–)-eudesmin and (+)-, (–)-yangambin. This method is useful to generate the core skeleton of furofuran rings by modification of asymmetric aldol condensation.

Acknowledgement

This work was supported by KOSEF (Grant No. RO5-2004-000-10742-0).

References and notes

- (a) Ward, R. S. *Chem. Soc. Rev.* **1982**, *11*, 75–125; (b) Ward, R. S. *Nat. Prod. Rep.* **1993**, *10*, 1–28; (c) Ward, R. S. *Nat. Prod. Rep.* **1995**, *12*, 183–205; (d) Ward, R. S. *Nat. Prod. Rep.* **1997**, *14*, 143–174; (e) Ward, R. S. *Nat. Prod. Rep.* **1999**, *16*, 75–96; (f) Whiting, D. A. *Nat. Prod. Rep.* **1990**, *7*, 349–364.
- (a) Teles, H. L.; Hemerly, J. P.; Pauletit, P. M.; Pandolfi, J. R. C.; Araujo, A. R.; Valentini, S. R.; Young, M. C. M.; Bolzani, V. S.; Silva, D. H. S. *Nat. Prod. Res.* **2005**, *19*, 319–323; (b) Chiung, Y. M.; Hayashi, H.; Matsumoto, H.; Otani, T.; Yoshida, K.; Huang, M. Y.; Chen, R. X.; Liu, J. R.; Nakayama, M. *J. Antibiot.* **1994**, *47*, 487–491.
- (a) De Leon, E. J.; Olmedo, D. A.; Solis, P. N.; Gupta, M. P.; Terencio, M. C. *Planta Med.* **2002**, *68*, 1128–1131; (b) Serra, M. F.; Diaz, B. L.; Barreto, E. O.; Pereira, A. P. B.; Lima, M. C. R.; Barbosa-Filho, J. M.; Cordeiro, R. S. B.; Martins, M. A.; De Silva, P. M. R. *Planta Med.* **1997**, *63*, 207–212.
- (a) Hoang, V. D.; Tan, G. T.; Zhang, H. J.; Tamez, P. A.; Nguyen, V. H.; Nguyen, M. C.; Soejarto, D. D.; Fong, H. H. S.; Pezzuto, J. M. *Phytochemistry* **2002**, *59*, 325–329; (b) Asano, J.; Chiba, K.; Tada, M.; Yoshi, T. *Phytochemistry* **1996**, *42*, 713–717.
- (a) MacRae, W. D.; Towers, G. H. N. *Phytochemistry* **1984**, *23*, 1207–1220; (b) Vermes, B.; Seligmann, O.; Wagner, H. *Phytochemistry* **1991**, *30*, 3087–3089.
- Brown, R. C.; Bataille, C. J.; Bruton, G.; Hinks, J. D.; Swain, N. A. *J. Org. Chem.* **2001**, *66*, 6719–6728.
- (a) Iwakami, S.; Ebizuka, Y.; Sankawa, U. *Heterocycles* **1990**, *30*, 795–798; (b) Iwakami, S.; Wu, J. B.; Ebizuka, Y.; Sankawa, U. *Chem. Pharm. Bull.* **1992**, *40*, 1196–1198.
- Ichikawa, K.; Kinoshita, T.; Nishibe, S.; Sankawa, U. *Chem. Pharm. Bull.* **1986**, *34*, 3514–3517.
- Nikaido, T.; Ohmoto, T.; Kinoshita, T.; Sankawa, U.; Nishibe, S.; Hisada, S. *Chem. Pharm. Bull.* **1981**, *29*, 3586–3592.
- Plante, G. E.; Prevost, C.; Chainey, A.; Braquet, P.; Sirois, P. *Am. J. Physiol.* **1987**, *253*, R375–R378.
- Bernard, C. B.; Arnason, J. T.; Philogene, B. J. R.; Lam, J.; Waddell, T. *Phytochemistry* **1989**, *28*, 1373–1377.
- (a) Brown, R. C. D.; Bataille, C. J. R.; Hinks, J. D. *Tetrahedron Lett.* **2001**, *42*, 473–475; (b) Van Oeveren, A.; Jansen, J. F. G. A.; Feringa, B. L. *J. Org. Chem.* **1994**, *59*, 5999–6007; (c) Ogiku, T.; Yoshida, S. I.; Ohmizu, H.; Iwasaki, T. *J. Org. Chem.* **1995**, *60*, 1148–1153; (d) Yoda, H.; Kimura, K.; Takabe, K. *Synlett* **2001**, 400–402; (e) Aldous, D. J.; Dalencon, A. J.; Steel, P. G. *Org. Lett.* **2002**, *4*, 1159–1162; (f) Roy, S. C.; Rana, K. K.; Guin, C. *J. Org. Chem.* **2002**, *67*, 3242–3248; (g) Swain, N. A.; Brown, R. C. D.; Bruton, G. *J. Org. Chem.* **2004**, *69*, 122–129; (h) Pelter, A.; Ward, R. S.; Watson, D. J.; Collins, P.; Kay, I. T. *J. Chem. Soc., Perkin Trans. 1* **1982**, 175–181; (i) Taylor, E. C.; Andrade, J. G.; Rall, G. J. H.; Steliou, K.; Jagdmann, G. E., Jr.; McKillop, A. *J. Org. Chem.* **1981**, *46*, 3078–3081; (j) Takano, S.; Ohkawa, T.; Tamori, S.; Satoh, S.; Ogasawara, K. *J. Chem. Soc., Chem. Commun.* **1988**, 189–191; (k) Orito, K.; Yorita, K.; Suginome, H. *Tetrahedron Lett.* **1991**, *32*, 5999–6002; (l) Suginome, H.; Ishikawa, M.; Yorita, K.; Shimoyama, N.; Sasaki, T.; Orito, K. *J. Org. Chem.* **1995**, *60*, 3052–3064; (m) Takeya, T.; Ara, Y.; Tobinaga, S. *Chem. Pharm. Bull.* **1995**, *43*, 1970–1976; (n) Samizu, K.; Ogasawara, K. *Chem. Lett.* **1995**, 543–544; (o) Hull, H. M.; Jones, R. G.; Knight, D. W. *J. Chem. Soc., Perkin Trans. 1* **1998**, 1779–1788; (p) Pelter, A.; Ward, R. S.; Watson, D. J.; Jack, I. R. *J. Chem. Soc., Perkin Trans. 1* **1982**, 183–190; (q) Beroza, M.; Schechter, M. S. *J. Am. Chem. Soc.* **1956**, *78*, 1242–1247; (r) Kise, N.; Fujimoto, A.; Ueda, N. *Tetrahedron: Asymmetry* **2002**, *13*, 1845–1847; (s) Aldous, D. J.; Dalencon, A. J.; Steel, P. G. *J. Org. Chem.* **2003**, *68*, 9159–9161; (t) Kasahara, H.; Miyazawa, M.; Kameoka, H. *Phytochemistry* **1997**, *44*, 1479–1482; (u) Banerjee, B.; Roy, S. C. *Synthesis* **2005**, 2913–2919.
- Kim, J. C.; Kim, K. H.; Jung, J. C.; Park, O. S. *Tetrahedron: Asymmetry* **2006**, *17*, 3–6.
- Gage, J. R.; Evans, D. A. *Org. Synth.* **1990**, *68*, 77–82.
- Jung, J. C.; Kache, R.; Vines, K. K.; Zheng, Y.; Bijoy, P.; Valluri, M.; Avery, M. A. *J. Org. Chem.* **2004**, *69*, 9269–9284.
- (a) Inoue, T.; Liu, J. F.; Buske, D. C.; Abiko, A. *J. Org. Chem.* **2002**, *67*, 5250–5256; (b) Evans, D. A.; Clark, J. S.; Metternich, R.; Novack, V. J.; Sheppard, G. S. *J. Am. Chem. Soc.* **1990**, *112*, 866–868.
- General procedure for the preparation of dilactones 15a and 16a:** To a stirred solution of substrates **9a** and **12** (2.2 mmol) in a specified solvent (12 mL) was added dropwise a specified enolizable reagent (4.8 mmol) and DIPEA (8.8 mmol, for boron aldol reagents) at –78 °C, and the reaction mixture was stirred at that temperature for 10 min. A solution of aldehydes (3,4-dimethoxybenzaldehyde, 2,3,4-trimethoxybenzaldehyde; 5.5 mmol) in solvent (THF or CH₂Cl₂; 2 mL) was added dropwise at –78 °C, and the resulting mixture was warmed to –40 or 0 °C for an appropriate reaction time (Table 1). The reaction mixture was quenched by the slow addition of satd aqueous NH₄Cl solution (5 mL). The mixture was warmed to room temperature, and diluted with dichloromethane (10 mL). The organic phase was separated, and the aqueous layer was extracted with dichloromethane (8 mL). The combined organic phases were dried over anhydrous MgSO₄, filtered, and concentrated under reduced pressure to give the aldol products **10a** and **11a**,

which were treated with KH_2PO_4 (10 mL), H_2O_2 (10 mL, 28%), MeOH (15 mL), and the resulting mixture was vigorously stirred at room temperature for 8 h. The mixture was extracted with dichloromethane (3×30 mL). The combined organic phase was dried over anhydrous MgSO_4 , filtered, and concentrated under reduced pressure to give dilactones, which were purified by flash column chromatography (silica gel, 25% ethyl acetate in hexanes) to afford pure dilactones **15a** and **16a**. Compound **15a**: Mp 203–204 °C (lit.,^{12h,18} 199 °C); IR (neat, NaCl) 3045, 2987, 1780, 1490, 1242, 1045, 1038 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 6.86 (s, 4H), 6.79 (s, 2H), 5.89 (s, 2H), 3.90 (s, 6H), 3.87 (s, 6H), 3.57 (s, 2H); ^{13}C NMR (100 MHz,

CDCl_3) δ 175.0, 149.6, 130.4, 116.7, 111.4, 107.9, 81.8, 56.1, 56.0, 48.4; HRMS calcd for $\text{C}_{22}\text{H}_{23}\text{O}_8$: 415.1393 $[\text{M}+\text{H}]^+$, found: 415.1398; **16a**: mp 170–171 °C; IR (neat, NaCl) 3030, 2975, 1778, 1600, 1520, 1261, 1045 cm^{-1} ; ^1H NMR¹²ⁱ (400 MHz, CDCl_3) δ 6.51 (s, 4H), 5.87 (s, 2H), 3.87 (s, 12H), 3.81 (s, 6H), 3.57 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 174.9, 153.9, 138.3, 133.6, 101.3, 81.6, 60.9, 56.3, 48.5; MS (ESI) (m/z) 475 $[\text{M}+\text{H}]^+$, 307, 154 (base peak); HRMS calcd for $\text{C}_{24}\text{H}_{27}\text{O}_{10}$: 475.1604 $[\text{M}+\text{H}]^+$, found: 475.1610.

18. Anjaneyulu, A. S. R.; Rao, A. M.; Rao, V. K.; Row, L. R.; Pelter, A.; Ward, R. S. *Tetrahedron* **1977**, *33*, 133–143.